

Naming compounds Practice page here—[answers on page 2](#)

barium chromate	
zinc dichromate	
cobalt (II) permanganate	
nickel (II) sulfite	
potassium phosphide	
	$\text{Cr}(\text{C}_2\text{H}_3\text{O}_2)_3$
	HgO
	H_2S
	H_2O
ammonia	
lead (IV) chlorite	
ammonium phosphate	
silver carbonate	
gold (III) fluoride	
copper (I) cyanide	
	Cl_2O_5
	CrI_6
	$\text{Al}_2(\text{C}_2\text{O}_4)_3$
	$\text{Ca}(\text{SCN})_2$
diarsenic trisulfide	
lead (II) chromate	
hydrogen fluoride	
sodium hydroxide	
cesium hydrogen carbonate	
magnesium hydrogen sulfate	

barium chromate	BaCrO_4
zinc dichromate	ZnCr_2O_7
cobalt (II) permanganate	$\text{Co}(\text{MnO}_4)_2$
nickel (II) sulfite	NiSO_3
potassium phosphide	K_3P
chromium (III) acetate	$\text{Cr}(\text{C}_2\text{H}_3\text{O}_2)_3$
mercury (II) oxide	HgO
hydrogen sulfide	H_2S
dihydrogen monoxide	H_2O
ammonia	NH_3
lead (IV) chlorite	$\text{Pb}(\text{ClO}_2)_4$
ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
silver carbonate	Ag_2CO_3
gold (III) fluoride	AuF_3
copper (I) cyanide	CuCN
dichlorine pentoxide	Cl_2O_5
chromium (VI) iodide	CrI_6
aluminum oxalate	$\text{Al}_2(\text{C}_2\text{O}_4)_3$
calcium thiocyanate	$\text{Ca}(\text{SCN})_2$
diarsenic trisulfide	As_2S_3
lead (II) chromate	PbCrO_4
hydrogen fluoride	HF
sodium hydroxide	NaOH
cesium hydrogen carbonate	CsHCO_3
magnesium hydrogen sulfate	$\text{Mg}(\text{HSO}_4)_2$