

REACTIONS HW #3 - Double Replacement Reactions name: _____

If you mix two aqueous solutions and a precipitate is formed, that is an indicator that a type of a double replacement reaction has occurred. BALANCE THESE EQUATIONS with phase symbols. Use table F to define AQ or S for each product. One "reaction" is not(?)

Sodium carbonate + zinc chlorate solutions react

Copper (II) sulfate + calcium chloride solutions react

Potassium hydroxide + lead (IV) nitrate solutions react

Silver hydrogen carbonate + Iron (II) bromide solutions react

Barium hydroxide + lithium sulfate solutions react

Ammonium phosphate + tin (II) acetate solutions react

Calcium chromate + sodium sulfide solutions react

Strontium acetate + lithium carbonate solutions react

Ammonium phosphate + rubidium nitrate solutions react